

Tetrahedron 57 (2001) 25-37

Total synthesis of 12(*R*)-HETE, 12(*S*)-HETE, ²H₂-12(*R*)-HETE and LTB₄ from racemic glycidol via hydrolytic kinetic resolution

A. Rodríguez,^a M. Nomen,^{a,†} B. W. Spur,^{a,b,*} J. J. Godfroid^c and T. H. Lee^b

^aDepartment of Cell Biology, University of Medicine and Dentistry of New Jersey, SOM, Stratford, NJ 08084, USA ^bDepartment of Respiratory Medicine and Allergy, King's College London, Guy's Hospital, London SE1 9RT, UK ^cLaboratorie de Pharmacochimie Moléculaire, Université Paris 7, Paris 75251, France

Received 30 June 2000; revised 16 October 2000; accepted 18 October 2000

Abstract—The total synthesis of 12(R)-HETE, 12(S)-HETE (Samuelsson's HETE), $[14,15-^{2}H_{2}]-12(R)$ -HETE and Leukotriene B₄ from racemic glycidol is described. The key steps are the hydrolytic kinetic resolution of racemic TES-glycidol with salen-Co catalyst and the selective oxidation of primary silyl ethers, in the presence of secondary ones, under Swern conditions to give a short entry to both enantiomers of 12-HETE and LTB₄. © 2000 Elsevier Science Ltd. All rights reserved.

The metabolism of arachidonic acid has attracted much attention over the years.¹ 12(R)-HETE (1) is formed via the cytochrome P-450 pathway and is present in high concentration in psoriasis lesions.² Its enantiomer, 12(S)-HETE (2), the major 12-lipoxygenase metabolite in platelets,³ has been found to play a central role in various stages of metastatic process in tumors and is therefore a potential target for an anticancer treatment. 12(S)-HETE (2) inhibits tumor cell adhesion to endothelial cells.⁴

LTB₄ (**3**), a metabolite of arachidonic acid via the 5-lipoxygenase pathway, is a potent chemotactic agent for human eosinophils and neutrophils and a modulator of inflammatory responses.⁵ Newer results demonstrated that LTB₄ (**3**) has high antiviral activity towards DNA viruses as well as retroviruses, including HIV-1 and HIV-2, comparable with antiviral drugs, such as Acylovir or Ganciclovir,⁶ opening new perspectives for **3** and metabolically stable LTB₄ analogs.⁷

In the interest of evaluating the biological and pharmacological properties of these compounds it was necessary to obtain sufficient quantities by chemical synthesis. The first synthesis of 12(S)-HETE (2) was reported by Corey et al.⁸ starting from S(-)-malic acid that was used later on as starting material by other groups.⁹ D(+)-Mannitol and

0040–4020/01/\$ - see front matter S 2000 Elsevier Science Ltd. All rights reserved. PII: S0040-4020(00)00983-2

D(-)-arabinose were used in the synthesis of 12(*S*)-HETE (2) whereas L(+)-arabinose has been the chiral source of 12(*R*)-HETE (1) and LTB₄ (3).¹⁰ Alpina-borane reduction¹¹ and the Sharpless kinetic resolution of racemic γ -iodoallylic alcohols (Sato's procedure)¹² were used in the synthesis of both enantiomers of 12-HETE and LTB₄. A recent report by Corey et al. described the synthesis of 12(*R*)-HETE and 12(*S*)-HETE utilizing the Sharpless asymmetric dihydroxylation.¹³

In this paper we report a short synthesis of the two enantiomers of 12-HETE, $[14,15-{}^{2}H_{2}]-12(R)$ -HETE (**37**) and LTB₄ (**3**) (Fig. 1) from easily available starting materials. The synthesis of 12(R)-HETE (**1**) and LTB₄ (**3**) were based on the retrosynthetic analysis shown in Fig. 2.

The chiral TES glycidol (7),¹⁴ a versatile C-3 building block, was obtained from racemic glycidol with >99% ee using Jacobsen's hydrolytic kinetic resolution with H₂O in the presence of 1% (*S*,*S*)-salen-Co catalyst (**14**) in Et₂O at room temperature,¹⁵ as shown in Scheme 1. Simple bulb to bulb distillation gave **7** in 46% isolated yield (92% theoretical yield).

The nucleophilic opening of the epoxide **7** with 1-heptynyllithium in the presence of TMEDA, similar as described by Nicolaou,¹⁶ was very slow. The protective group was cleaved under those conditions and large quantities of the dimerization product of 1-heptyne were isolated. Better results were achieved using the Yamaguchi method with $BF_3 \cdot Et_2O$ at $-78^{\circ}C$ in THF.¹⁷ The reaction was fast and only a small amount of deprotected material was observed. The optical purity was measured

Keywords: eicosanoids; hydrolytic kinetic resolution; selective Swern oxidation.

^{*} Corresponding author. Department of Cell Biology, University of Medicine and Dentistry of New Jersey, SOM, Stratford, NJ 08084, USA. Tel.: +1-856-566-7016; fax: +1-856-566-6195; e-mail: spurbw@umdnj.edu

[†] Deceased on 26 September 1997.

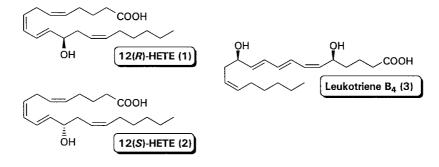
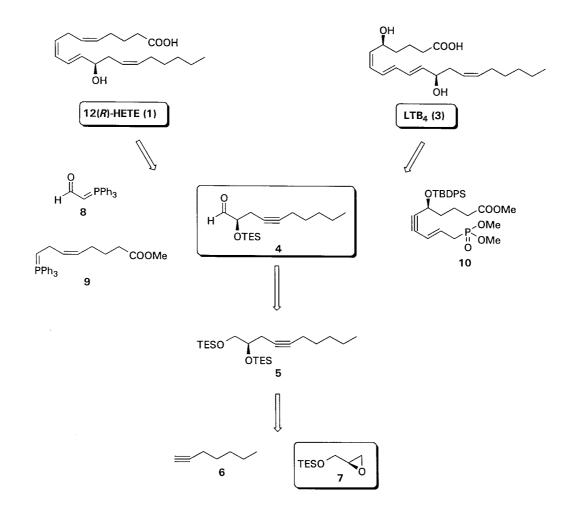


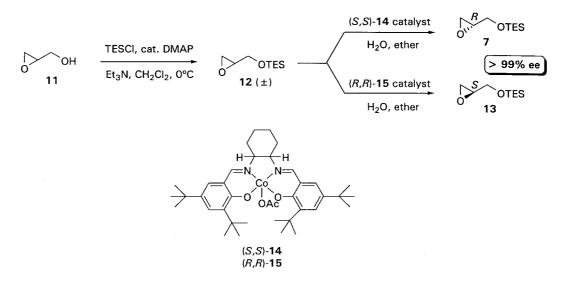
Figure 1.

by chiral HPLC [Column: OF-chiracel, mobile phase: hexane/*i*-PrOH (90:10), λ =265 nm, >99% ee] of the dibenzoate derivative obtained from compound **16** after cleavage of the TES ether. The secondary hydroxy group in **16** was protected with triethylsilylchloride to give **5**. Selective Swern oxidation of the primary TES ether in the presence of the secondary one produced directly the chiral key intermediate **4** (Scheme 2), required for the synthesis of 12(*R*)-HETE (**1**) and LTB₄ (**3**).¹⁵

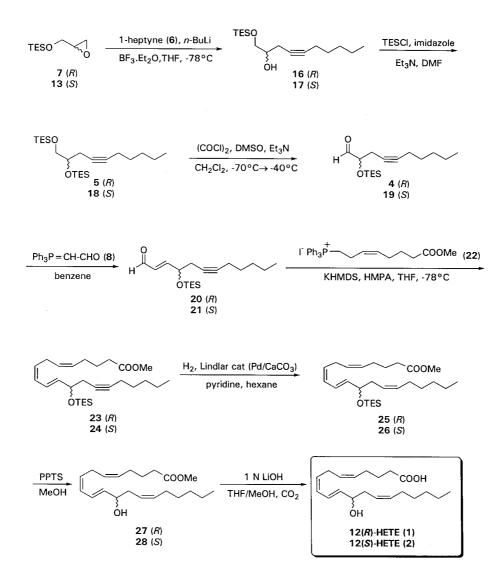
The synthesis of 12(R)-HETE (1) (Scheme 2) was accomplished, similarly as described by Rokach et al., ^{10f} by Wittig reaction of the aldehyde **4** with 3 equivalents of (triphenyl-

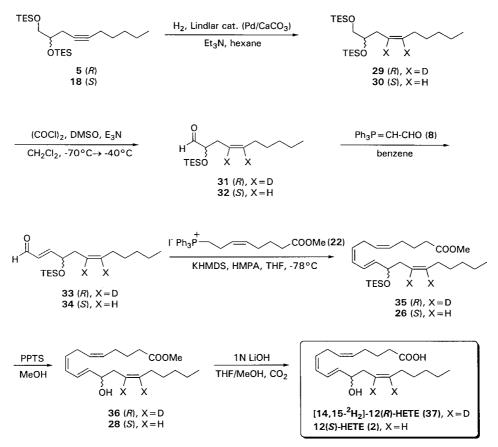
phosphoranylidene)acetaldehyde (8)¹⁸ in benzene at 60°C to produce the α , β -unsaturated aldehyde 20 in 87% isolated yield. Compound 20 was reacted with methyl 8-(triphenylphosphoranylidene)-5(*Z*)-octenoate (9), generated in situ from 22 with KHMDS, in THF/HMPA at -78° C to give the 12(*R*)-HETE precursor 23. The critical step in the synthesis was the selective reduction of the triple bond in 23 to the *cis* double bond in 25 in the presence of the additional conjugated and isolated double bonds avoiding over-reduction.¹⁹ Reaction conditions and solvents were very critical. Compound 25 was obtained from 23 by controlled hydrogenation (Lindlar catalyst, deactivated with pyridine, in hexane). 12(*R*)-HETE methyl ester (27)





Scheme 1.





Scheme 3.

was obtained in high yield by mild deprotection of the silyl ether in **25** with pyridinium *p*-toluenesulfonate in acetone/ H_2O (2 h) or in MeOH (1.5 h). Compound **27** was characterized by ¹H NMR, ¹³C NMR, APT, COSY, HETCOR and optical rotation. Final hydrolysis of **27** with a 1N aqueous solution of LiOH in THF/MeOH gave 12(*R*)-HETE (1) identical in all aspects with an authentic sample (Upjohn-Pharmacia).

The synthesis of the major platelet metabolite of arachidonic acid, 12(S)-HETE (2), was achieved using a similar route as described for its enantiomer 12(R)-HETE (1) (Scheme 2). The chiral TES-glycidol (13) was obtained with >99% ee using Jacobsen's hydrolytic kinetic resolution with 1% (*R*,*R*)-salen-Co catalyst (15) (Scheme 1). Compound 2 was characterized by ¹H NMR, ¹³C NMR, APT, COSY, HETCOR and HPLC/API-ES/MS, and was identical in all aspects with an authentic sample (Upjohn-Pharmacia).

In order to avoid the difficulties encountered in the Lindlar reduction at a late stage we examined the reduction of the triple bond to the *cis* double bond at the first possible step. The synthesis of $[14,15-{}^{2}H_{2}]-12(R)$ -HETE (**37**) and 12(S)-HETE (**2**) is outlined in Scheme 3. The di-TES ether **5** was cleanly reduced in hexane with deuterium, Lindlar catalyst and Et₃N affording **29** in quantitative yield. In a similar way intermediate **18** was reduced with hydrogen affording **30**.

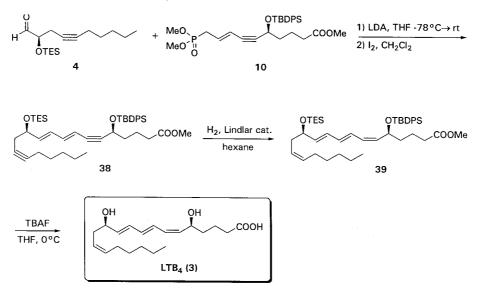
Compound **29** was transformed to $[14,15^{-2}H_2]^{-12}(R)^{-12}$

(37), as outlined in Scheme 3, through intermediates 31 and 33 to give $[14,15^{-2}H_2]-12(R)$ -HETE methyl ester (36). Mild hydrolysis as previously described gave $[14,15^{-2}H_2]-12(R)$ -HETE (37).

Compound **30** was transformed to **2** following the same steps as described for **37**, as outlined in Scheme 3. Compound **2** was identical with the material prepared by the method described in Scheme 2.

The synthesis of LTB₄ (**3**) was accomplished by Emmons– Horner reaction of aldehyde **4** with Nicolaou's chiral C_1-C_{10} phosphonate **10**.¹⁶ (Scheme 4) The reaction of **4** with **10** produced unexpectedly the 8*E*,10*Z*-diene, however, the crude product could be easily isomerized with a catalytic amount of iodine in CH₂Cl₂ at room temperature to give **38** (>95% *E,E*). Lindlar hydrogenation of **38** followed by deprotection with TBAF gave LTB₄ (**3**) directly, identical in all aspects (¹H NMR, ¹³C NMR, APT, COSY, HETCOR, UV, optical rotation, HPLC, HPLC/API-ES/MS and bioassay) with a sample prepared using Corey's route.²⁰

In summary, a short and efficient total synthesis of 12(S)-HETE, 12(R)-HETE, $[14,15^{-2}H_2]-12(R)$ -HETE and LTB₄ has been accomplished from racemic glycidol. The key steps are the Jacobsen hydrolytic kinetic resolution of racemic TES-glycidol and the selective oxidation of the primary TES ether in the presence of a secondary one. The application of this method towards other natural products will be reported in due course.



Scheme 4.

1. Experimental

1.1. General remarks

All reactions that were moisture and air-sensitive were carried out in flame-dried glassware and under an argon atmosphere. The progress of the reactions was checked by thin layer chromatography (TLC) using E. Merck silica gel 60F glass plates (0.25 mm). The spots were visualized with UV light, followed by heat staining with p-anisaldehyde in EtOH/CH₃COOH/H₂SO₄. Silica gel 60 from EM-Science was used for flash chromatography. HPLC analysis were performed on a Hewlett-Packard liquid chromatograph HP-1090 Series II with PV5 SDS (Solvent Delivery System) and DAD (Diode Array Detector) equipped with heated column compartment and automatic liquid injector, or on a Waters HPLC system (M-6000A pump, M-730 Data Module integrator, U6K Injector) and a Schoeffel SF-770 UV detector. For Chiral HPLC Chiracel OF, OD, OB columns (Daicel Chemical Ltd.) have been used. ¹H NMR and ¹³C NMR data were recorded on a 300 MHz Varian Gemini 2000 Broadband High-Resolution NMR. IR spectra were measured on a Perkin-Elmer Paragon 1000 FT-IR spectrometer. UV Spectra were obtained using a Hewlett Packard HP-8453 UV-Visible Spectrophotometer. Optical rotation was measured on a Perkin-Elmer Polarimeter 343. Mass Spectra were obtained using Hewlett Packard HP-59987A API-Electrospray (Atmospheric Pressure Ionization Electrospray) interface coupled to a Mass Spectrometer Hewlett Packard HP-5989B MS. High Resolution Mass Spectra (HRMS) were obtained at the University of Pennsylvania Mass Spectrometry Service Center on a VG Micromass using Electrospray Ionization Mode. Microanalysis were performed by Micro-Analysis, Inc., Wilmington, Delaware.

1.1.1. (Triethylsilyl)glycidyl ether (12). To a solution of glycidol (11) (7.4 g, 0.10 mol), Et_3N (15.9 ml, 0.11 mol) and 4-dimethylaminopyridine (50 mg, 0.41 mmol) in CH_2Cl_2 (50 ml) at 0°C was added dropwise chlorotriethyl-silane (15.9 ml, 94.7 mmol). The reaction was allowed to

reach room temperature and stirred until completion. The mixture was diluted with CH_2Cl_2 and poured in ice-cold H_2O . Separation of the organic phase, washing with a saturated solution of NaCl, drying over Na₂SO₄ and concentration under vacuo afforded crude **12** in quantitative yield. The product was purified by bulb to bulb distillation 60–80°C/1 mm affording **12** (15.7 g, 84%). ¹H NMR (CDCl₃, 300 MHz): δ 3.8 (1H, ddd, *J*=11.8, 3.3, 0.4 Hz, ¹H of CH₂–OTES), 3.6 (1H, dd, *J*=11.8, 5.0 Hz, ¹H of CH₂–OTES), 3.6 (1H, dd, *J*=5.1, 2.8 Hz, ¹H of CH₂–O-CH), 0.9 (9H, t, *J*=7.8 Hz, CH₃–CH₂–Si), 0.6 (6H, q, *J*=7.8 Hz, CH₂–Si). ¹³C NMR (CDCl₃, 75.5 MHz): δ 63.6 (CH₂–OTES), 52.3 (CH–O), 44.5 (CH₂–O–CH), 6.5 (3C, CH₃–CH₂–Si), 4.2 (3C, CH₂–Si).

1.1.2. (R)-(Triethylsilyl)glycidyl ether (7). Compound 12 (12.2 g, 64.8 mmol) was dissolved in Et₂O (25 ml) and (S,S)-(salen)Co(III)(AcO) (14) (0.21 g, 0.32 mmol) was added. The mixture was cooled to 0°C and H₂O (0.64 ml, 35.6 mol) was added dropwise. The reaction was allowed to reach room temperature and stirred for 20 h. The Et₂O was evaporated and the residue was purified by bulb to bulb distillation 50–55°C/0.6 mm affording 7 (5.6 g, 46% yield). $[\alpha]_D^{25}=1.9$ (c=1.08, CHCl₃). ¹H NMR (CDCl₃, 300 MHz): δ 3.8 (1H, ddd, J=11.8, 3.3, 0.4 Hz, ¹H of CH₂-OTES), 3.6 (1H, dd, J=11.8, 5.0 Hz, ¹H of CH₂-OTES), 3.1 (1H, m, CH-O), 2.8 (1H, ddd, J=5.1, 4.0, 0.4 Hz, ¹H of CH₂-O-CH), 2.6 (1H, dd, J=5.1, 2.8 Hz, ¹H of CH₂–O–CH), 0.9 (9H, t, J=7.8 Hz, CH₃–CH₂–Si), 0.6 (6H, q, J=7.8 Hz, CH₂-Si). ¹³C NMR (CDCl₃, 75.5 MHz): δ 63.6 (CH₂-OTES), 52.3 (CH-O), 44.5 (CH₂-O-CH), 6.5 (3C, CH₃-CH₂-Si), 4.3 (3C, CH₂-Si).

1.1.3. (*R*)-1-[(Triethylsilyl)oxy]-4-decyn-2-ol (16). A solution of 1-heptyne (3.5 ml, 26.8 mmol) in THF (20 ml) at -78° C was treated with a 2.5 M solution of *n*-butyllithium in hexane (10.7 ml, 26.8 mmol). The solution was brought to 0°C and kept at that temperature for 15 min. The reaction was cooled again to -78° C and then the TES protected glycidol 7 (2.4 g, 12.7 mmol) in THF (5 ml) was added

followed by BF₃·Et₂O (3.4 ml, 26.8 mmol). After 30 min stirring the reaction was quenched by the addition of a saturated solution of NH₄Cl. The reaction was brought to room temperature and diluted with Et₂O. The phases were separated and the organic layer was washed with brine, dried over Na₂SO₄ and concentrated under vacuo. The crude product 16, that contained small amount of the unprotected diol, was used without further purification for the next step. A small sample was purified by flash chromatography for analytical purpose. ¹H NMR (CDCl₃, 300 MHz): δ 3.8–3.7 (2H, m, CH–OH, ¹H of CH₂–OTES), 3.7–3.5 (1H, m, ¹H of CH2-OTES), 2.6-2.5 (1H, br. s, OH), 2.4-2.3 (2H, m, CH2-CH-OH), 2.2-2.1 [2H, tt, J=7.0, 2.4 Hz, CH2-(CH₂)₃-CH₃], 1.5-1.4 [2H, m, CH₂-(CH₂)₂-CH₃], 1.4-1.2 [4H, m, $(CH_2)_2$ -CH₃], 1.0-0.9 (9H, t, J=8.0 Hz, CH₃-CH₂-Si), 0.9-0.8 (3H, t, J=7.2 Hz, CH₃), 0.6 (6H, q, J=8.0 Hz, CH₂-Si). ¹³C NMR (CDCl₃, 75.5 MHz): δ 82.6 (\equiv C), 75.6 (\equiv C), 70.5 (CH–OH), 65.4 (CH₂– OTES), 31.0 (CH₂-CH₂-CH₃), 28.6 [CH₂-(CH₂)₂-CH₃], 23.3 (CH₂-CH-OH), 22.1 (CH₂-CH₃), 18.6 [CH₂-(CH₂)₃-CH₃], 13.8 (CH₃), 6.5 (3C, CH₃-CH₂Si), 4.2 (3C, CH₂–Si). Enantiomeric excess was determined to be >99%by chiral HPLC of the dibenzoate derivative obtained from compound 16 after cleavage of the TES ether [Column: OF-chiracel, mobile phase: hexane/i-PrOH (90:10), $\lambda = 265$ nm].

1.1.4. (R)-1,2-Bis[(triethylsilyl)oxy]-4-decyne (5). To a solution of mono-protected compound 16 (3.6 g, 12.7 mmol) and imidazole (4.6 g, 67.6 mmol) in DMF (15 ml) at 0° C was added Et₃N (5.1 ml, 36.6 mmol) followed by chlorotriethylsilane (4.5 ml, 26.7 mmol). After 30 min stirring, the reaction was quenched over icecold H₂O layered with hexane/EtOAc (9:1). The organic layer was separated, washed four times with ice-cold H₂O and once with brine. After drying and concentrating under vacuo the crude was purified by flash chromatography with hexane/EtOAc (98:2) (0.5% Et₃N) and 5 (3.2 g, 64% twosteps) was obtained. $[\alpha]_D^{25}=2.0$ (c=1, CH₂Cl₂). ¹H NMR (CDCl₃, 300 MHz): δ 3.8 (1H, m, CH-OTES), 3.7-3.5 (2H, 2 dd AB system, J=9.9, 5.7 Hz, CH₂-OTES), 2.5-2.4 (1H, ddt AB system, J=16.5, 6.0, 2.4 Hz, ¹H of CH_2 -CH-OTES), 2.3-2.2 (1H, ddt AB system, J=16.5, 5.7, 2.4 Hz, ¹H of CH_2 -CH-OTES), 2.1 [2H, tt, J=6.9, 2.4 Hz, CH₂-(CH₂)₃-CH₃], 1.5-1.4 [2H, m, CH₂- $(CH_2)_2-CH_3$, 1.4–1.2 [4H, m, $(CH_2)_2-CH_3$], 1.0–0.8 (21H, m, CH₃-CH₂-Si, CH₃), 0.7-0.5 (12H, m, CH₂-Si). ¹³C NMR (CDCl₃, 75.5 MHz): δ 81.6 (\equiv C), 77.0 (\equiv C), 72.5 (CH-OTES), 66.4 (CH2-OTES), 31.1 (CH2-CH2-CH₃), 28.7 [CH₂-(CH₂)₂-CH₃], 24.6 (CH₂-CH-OTES), 22.2 (CH₂-CH₃), 18.7 [CH₂-(CH₂)₃-CH₃], 13.8 (CH₃), 6.7 (3C, CH₃-CH₂-Si), 6.6 (3C, CH₃-CH₂-Si), 4.9 (3C, CH₂-Si), 4.3 (3C, CH₂-Si). IR (film): 2955, 2935, 2912, 2875, 1459, 1415, 1239, 1123, 1081, 1005, 815, 743, 672 cm^{-1} .

1.1.5. (*R*)-2-[(Triethylsilyl)oxy]-4-decynal (4). To a -70° C solution of oxalyl chloride (0.19 ml, 2.2 mmol) in CH₂Cl₂ (2 ml) was added dropwise DMSO (0.31 ml, 4.4 mmol). After 15 min stirring the di-TES ether **5** (0.20 g, 0.5 mmol) in CH₂Cl₂ (2 ml) was added. The stirring was continued at -70° C for 20 min followed by 20 min at -40° C. The reaction was quenched at -70° C by the addi-

tion of Et₃N (1.05 ml, 7.5 mmol). The mixture was allowed to reach room temperature, diluted with CH₂Cl₂ and washed with ice-cold H₂O and brine. After drying and concentrating the crude was purified by flash chromatography affording aldehyde **4** (96 mg, 68%). ¹H NMR (CDCl₃, 300 MHz): δ 9.6 (1H, d, *J*=1.8 Hz, CHO), 4.0 (1H, m, CH–OTES), 2.6–2.4 (2H, m, CH₂–CH–OTES), 2.2–2.0 [2H, m, CH₂–(CH₂)₃–CH₃], 1.5–1.4 [2H, m, CH₂–(CH₂)₂–CH₃], 1.4–1.2 [4H, m, (CH₂)₂–CH₃], 1.0–0.8 (12H, m, CH₃, CH₃–CH₂–Si), 0.6 (6H, m, CH₂–Si). ¹³C NMR (CDCl₃, 75.5 MHz): δ 202.2 (CHO), 83.0 (=C), 76.0 (CH–OTES), 74.4 (=C), 30.9 (CH₂–CH₂–CH₃), 28.3 [CH₂–(CH₂)₂–CH₃], 23.4 (CH₂–CH–OTES), 22.0 (CH₂–CH₃), 18.5 [CH₂–(CH₂)₃–CH₃], 13.7 (CH₃), 6.4 (3C, CH₃–CH₂–Si), 4.6 (3C, CH₂–Si). IR (film): 2955, 2933, 2875, 1741, 1459, 1239, 1122, 1077, 1007, 744, 730 cm⁻¹.

1.1.6. (R)-4-[(Triethylsilyl)oxy]-2(E)-dodecen-6-ynal (20). To the aldehyde 4 (96 mg, 0.34 mmol) in benzene (5 ml) was added (triphenylphosphoranylidene)acetaldehyde (8) (0.30 g, 1.0 mmol). The reaction mixture was brought to 60°C and stirred for 2 h. The solvent was evaporated under vacuo and the crude product was purified by flash chromatography [SiO₂, hexane/EtOAc (98:2) (1% Et₃N)] affording 20 (91 mg, 87%). $[\alpha]_D^{25} = -45.3$ (c=1.03, CH₂Cl₂). ¹H NMR (CDCl₃, 300 MHz): δ 9.6 (1H, d, J=7.8 Hz, CHO), 6.9 (1H, dd, J=15.3, 4.2 Hz, =CH-CH-OTES), 6.3 (1H, ddd, J=15.3, 7.8, 1.5 Hz, =CH-CHO), 4.5 (1H, dddd, J=8.4, 5.7, 4.2, 1.5 Hz, CH-OTES), 2.6-2.5 (1H, ddt AB system, J=16.5, 5.7, 2.4 Hz, ¹H of CH₂-CH-OTES), 2.4-2.3 (1H, ddt AB system, J=16.5, 8.4, 2.4 Hz, ¹H of CH₂-CH-OTES), 2.1 [2H, tt, J=7.2, 2.4 Hz, $CH_2-(CH_2)_3-CH_3$], 1.5–1.4 [2H, m, CH₂-(CH₂)₂-CH₃], 1.4-1.2 [4H, m, (CH₂)₂-CH₃], 1.0-0.8 (12H, m, CH₃, CH₃-CH₂-Si), 0.6 (6H, m, CH₂-Si). ¹³C NMR (CDCl₃, 75.5 MHz): δ 193.6 (CHO), 158.3 (=CH-CH-OTES), 131.2 (=CH-CHO), 83.4 (≡C), 75.1 (≡C), 70.8 (CH-OTES), 31.0 (CH2-CH2-CH3), 28.4 [CH2-(CH₂)₂-CH₃], 28.0 (CH₂-CH-OTES), 22.1 (CH₂-CH₃), 18.6 [CH₂-(CH₂)₃-CH₃], 13.8 (CH₃), 6.6 (3C, CH₃-CH₂-Si), 4.6 (3C, CH₂-Si). IR (film): 2956, 2934, 2915, 2876, 2808, 2719, 1696, 1459, 1107, 1006, 975, 744, 729 cm⁻¹.

1.1.7. (*R*)-12-[(Triethylsilyl)oxy]-5(*Z*),8(*Z*),10(*E*)-eicosatrien-14-ynoic acid methyl ester (23). To a suspension of the phosphonium iodide salt 22 (0.20 g, 0.40 mmol) in THF (1.5 ml) at -78° C was added dropwise a 0.5 M solution of potassium bis(trimethylsilyl)amide in toluene (0.84 ml, 0.42 mmol). The mixture was stirred for 30 min and aldehyde 20 (72.0 mg, 0.23 mmol) in THF (1.5 ml) followed by HMPA (0.25 ml) were added. After 30 min stirring at -78°C the mixture was brought to 0°C and added to a saturated solution of NH₄Cl layered with hexane/EtOAc (9:1). The phases were separated and the organic layer was washed with brine. After drying over Na₂SO₄ and concentrating under vacuo the crude was purified by flash chromatography affording **23** (88 mg, 86%). $[\alpha]_{\rm D}^{25} = -19.2$ (c=0.83, acetone). ¹H NMR (d₆-benzene, 300 MHz): δ 6.9 (1H, ddt, J=15.2, 11.1, 1.2 Hz, CH=CH-CH-OTES), 6.2–6.1 (1H, br. t, J=11.1 Hz, =CH-CH=CH-CH-OTES), 6.0 (1H, dd, J=15.2, 6.0 Hz, =CH-CH-OTES), 5.6–5.3 [3H, m, $=CH-CH_2-CH=CH-(CH_2)_3-COO$], 4.5 (1H, m, CH-OTES), 3.4 (3H, s, CH₃O), 3.0 [2H, br. t, J=7.4 Hz, CH₂-CH=CH-(CH₂)₃-COO], 2.7-2.6 (1H, ddt AB system, J=16.2, 6.0, 2.4 Hz, ¹H of CH₂-CH-OTES), 2.6–2.5 (1H, ddt AB system, J=16.2, 7.0, 2.4 Hz, ¹H of CH₂-CH-OTES), 2.2 [4H, m, CH₂-(CH₂)₃-CH₃, CH₂-COO], 2.0 [2H, m, CH_2 -(CH₂)₂-COO], 1.8-1.6 (2H, quint, J=7.2 Hz, CH₂-CH₂-COO), 1.6-1.5 [2H, m, CH₂-(CH₂)-COO], 1.5-1.2 [4H, m, (CH₂)₂-CH₃], 1.1 (9H, t, J=8.0 Hz, CH₃-CH₂-Si), 0.9 (3H, t, J=7.9 Hz, CH₃), 0.7 (6H, q, J=7.9 Hz, CH₂-Si). ¹³C NMR (d₆-benzene, 75.5 MHz): δ 173.2 (COO), 136.5 (=CH-CH-OTES), 130.1 (=CH), 129.6 [=CH-(CH₂)₃-COO], 128.7 128.6 (=CH-CH=CH-CH-OTES),125.3 (=CH), (CH=CH-CH-OTES), 82.3 (≡C), 77.1 (≡C), 72.6 (CH-OTES), 50.7 (CH₃-O), 33.2 (CH₂-COO), 31.2 (CH2-CH2-CH3), 29.4 (CH2-CH-OTES), 28.9 [CH2-(CH₂)₂-CH₃], 26.6 [CH₂-(CH₂)₂-COO], 26.2 (=CH-CH₂-CH=), 24.9 (CH₂-CH₂-COO), 22.3 (CH₂-CH₃), 19.0 $[CH_2-(CH_2)_3-CH_3]$, 13.9 (CH_3) , 6.9 $(3C, CH_3-$ CH₂-Si), 5.2 (3C, CH₂-Si). IR (film): 3011, 2953, 2933, 2875, 1741, 1458, 1435, 1240, 1155, 1107, 1073, 1005, 984, 951, 743, 727 cm⁻¹. HRMS (ESI) calcd for C₂₇H₄₆O₃SiNa $(M+Na)^+$: 469.3114, found $(M+Na)^+$: 469.3106.

1.1.8. (R)-12-[(Triethylsilyl)oxy]-5(Z),8(Z),10(E),14(Z)eicosatetraenoic acid methyl ester (25). To a solution of alkyne 23 (88 mg, 0.20 mmol) in hexane (5 ml) under argon were added pyridine (92 μ L) and Lindlar catalyst (130 mg). The argon atmosphere was exchanged by hydrogen and the mixture was stirred for 36 h. Filtration of the catalyst, evaporation of the solvent and purification by flash chromatography with hexane/EtOAc (95:5) (1% Et₃N) afforded 25 (75 mg, 85%). ¹H NMR (CDCl₃, 300 MHz): δ 6.4 (1H, ddt, J=15.0, 11.1, 1.2 Hz, CH=CH-CH-OTES), 6.0 (1H, br. t, J=11.1 Hz, =CH-CH=CH-CH-OTES), 5.7-5.6 (1H, dd, J=15.0, 6.3 Hz, =CH-CH-OTES), 5.5-5.2 [5H, m, =CH-CH₂-CH=CH-(CH₂)₃-COO, CH=CH-(CH₂)₄-CH₃], 4.2 (1H, m, CH-OTES), 3.6 (3H, s, CH₃O), 2.9 (2H, m, =CH-CH₂-CH=), 2.3 (2H, t, J=7.5 Hz, CH₂-COO), 2.3-2.2 (2H, m, CH₂-CH-OTES), 2.1 [2H, m, $CH_2-(CH_2)_2-COO]$, 2.0 [2H, m, $CH_2-(CH_2)_3-CH_3$], 1.7 (2H, quint, J=7.5 Hz, CH_2-CH_2-COO), 1.4–1.2 [6H, m, (CH₂)₃-CH₃], 1.0-0.9 (9H, m, CH₃-CH₂-Si), 0.9 (3H, t, J=6.9 Hz, CH₃), 0.6 (6H, m, CH₂-Si). ¹³C NMR (CDCl₃, 75.5 MHz): δ 174.1 (COO), 137.0 (=*C*H-CH-OTES), 132.1 [=CH-(CH₂)₄-CH₃], 129.5 (CH=CH-CH=CH-129.3 $[=CH-(CH_2)_3-COO],$ CH-OTES), 128.7 [CH=CH-(CH₂)₃-COO], 128.4 (=CH-CH=CH-CH-OTES), 125.2 [CH=CH-(CH₂)₄-CH₃], 124.5 (CH=CH-CH-OTES), 73.1 (CH-OTES), 51.3 (CH₃-O), 36.5 (CH₂-CH-OTES), 33.4 (CH₂-COO), 31.5 (CH₂-CH₂-CH₃), 29.2 [CH₂-(CH₂)₂-CH₃], 27.4 [CH₂-(CH₂)₃-CH₃], 26.5 $[CH_2-(CH_2)_2-COO], 26.0 (=CH-CH_2-CH=), 24.8$ (CH₂-CH₂-COO), 22.5 (CH₂-CH₃), 13.9 (CH₃), 6.7 (3C, CH₃-CH₂-Si), 5.0 (3C, CH₂-Si). IR (film): 3011, 2953, 2929, 2874, 1741, 1456, 1239, 1071, 1005, 743, 727 cm^{-1} . Anal. Calcd for $C_{27}H_{48}O_3Si$: C, 72.26; H, 10.78. Found: C, 72.69; H, 10.90.

1.1.9. 12(*R*)**-HETE methyl ester** (**27**)**.** To compound **25** (75 mg, 0.17 mmol) in MeOH (10 ml) was added pyridinium *p*-toluenesulfonate (15 mg). The mixture was stirred for 1.5 h at room temperature and then quenched by the addition of Et_3N . The solvent was evaporated under vacuo and crude **27** was purified by flash chromatography with

hexane/EtOAc (9:1) affording 12(R)-HETE methyl ester (47 mg, 84%). $[\alpha]_{\rm D}^{25} = -12$ (c=0.9, acetone). {Ref.^{12d} $[\alpha]_{D}^{20} = -11.4$ (c=1.0, acetone)}. UV (CH₃CN) λ_{max} 237 nm. ¹H NMR (CDCl₃, 300 MHz): δ 6.6–6.5 (1H, ddt, J=15.3, 11.1, 1.2 Hz, CH=CH-CH-OH), 6.0 (1H, br. t, J=11.1 Hz, =CH-CH=CH-CH-OH), 5.7 (1H, dd,J=15.3, 6.0 Hz, =CH-CH-OH), 5.6-5.5 [1H, m, =CH-(CH₂)₄-CH₃], 5.4-5.3 [4H, m, =CH-CH₂-CH=CH-(CH₂)₃-COO, CH=CH-(CH₂)₄-CH₃], 4.2 (1H, m, CH-OH), 3.6 (3H, s, CH₃O), 2.9 (2H, m, =CH-CH₂-CH=), 2.4-2.2 (4H, t and m, J=7.5 Hz, CH₂-COO, CH₂-CH-OH), 2.2-2.0 [4H, m, CH₂-(CH₂)₂-COO, CH₂-(CH₂)₃-CH3], 2.0-1.9 (1H, br. s, OH), 1.8-1.6 (2H, m, CH2-CH2-COO), 1.4-1.2 [6H, m, (CH2)3-CH3], 0.9 (3H, t, J=6.8 Hz, CH₃). ¹³C NMR (CDCl₃, 75.5 MHz): δ 174.2 (COO), 136.0 (=CH-CH-OH), 133.7 [=CH-(CH₂)₄-CH₃], 130.3, 129.4, 128.5 [=CH-CH₂-CH=CH-(CH₂)₃-COO], 128.1 (=CH-CH=CH-CH-OH), 125.5 (*C*H=CH-CH-OH), 124.4 $[CH = CH - (CH_2)_4 - CH_3],$ 72.1 (CH-OH), 51.4 (CH₃-O), 35.4 (CH₂-CH-OH), 33.4 (CH₂-COO), 31.5 (CH₂-CH₂-CH₃), 29.2 [CH₂-(CH₂)₂-CH₃], 27.4 [CH₂-(CH₂)₃-CH₃], 26.6 [CH₂-(CH₂)₂-COO], 26.1 (=CH-CH₂-CH=), 24.7 (CH₂-CH₂-COO), 22.5 (CH₂-CH₃), 13.9 (CH₃).

1.1.10. 12(R)-HETE (1). To a solution of 27 (47 mg, 0.14 mmol) in THF (10 ml) at 0°C under an argon atmosphere was added 1N LiOH (2.8 ml, 2.8 mmol). MeOH (1.2 ml) was added to achieve a homogeneous solution. The mixture was stirred for 2.5 h and quenched by the addition of solid carbon dioxide. The mixture was concentrated under vacuo, diluted with EtOAc and layered with phosphate buffer pH 5.6. The phases were separated, the aqueous layer was re-extracted with EtOAc and the combined organic layers were dried (Na₂SO₄) and concentrated under vacuo affording 12(R)-HETE (1) (38 mg, 85%). $[\alpha]_{\rm D}^{25} = -11.7$ (c=0.77, acetone). UV (CH₃CN) $\lambda_{\rm max}$ 237 nm. ¹H NMR (CDCl₃, 300 MHz): δ 6.6 (1H, ddt, J=15.3, 11.1, 1.2 Hz, CH=CH-CH-OH), 6.0 (1H, br. t, J=11.1 Hz, =CH-CH=CH-CH-OH), 5.7 (1H, dd, J=15.3, 6.3 Hz, =CH-CH-OH), 5.6-5.5 [1H, dtt, $J=11.1, 7.2, 1.5 \text{ Hz}, =CH-(CH_2)_4-CH_3], 5.5-5.3 [4H,$ $=CH-CH_2-CH=CH-(CH_2)_3-COO,$ CH=CHm, (CH₂)₄-CH₃], 4.3-4.2 (1H, m, CH-OH), 3.0-2.8 (2H, m, =CH-CH $_2-$ CH=), 2.4-2.3 (4H, m, CH $_2-$ COO, CH $_2-$ CH–OH), 2.1 [2H, m, CH₂–(CH₂)₂–COO], 2.1–2.0 [2H, m, CH₂-(CH₂)₃-CH₃], 1.7 (2H, m, CH₂-CH₂-COO), 1.4-1.2 [6H, m, (CH₂)₃-CH₃], 0.9-0.8 (3H, t, J=6.8 Hz, CH₃). ¹³C NMR (CDCl₃, 75.5 MHz): δ 178.1 (COO), 135.4 (=*C*H-CH-OH), 133.7 [=*C*H-(CH₂)₄-CH₃], 130.3, 129.4, 128.4 $[=CH-CH_2-CH=CH-(CH_2)_3-COO],$ 127.8 (=CH-CH=CH-CH-OH), 125.6 (CH=CH-CH-OH), 124.3 [CH=CH-(CH₂)₄-CH₃], 72.1 (CH-OH), 35.3 (CH₂-CH-OH), 33.1 (CH₂-COO), 31.4 (CH₂-CH₂-CH₃), 29.2 [CH₂-(CH₂)₂-CH₃], 27.4 [CH₂-(CH₂)₃-CH₃], 26.4 $[CH_2-(CH_2)_2-COO], 26.1 (=CH-CH_2-CH=), 24.5$ (CH₂-CH₂-COO), 22.4 (CH₂-CH₃), 13.9 (CH₃). HPLC/ API-ES/MS (m/z): 319 [$M_{12(R)-HETE}-H^+$]⁻.

1.1.11. (*S*)-(Triethylsilyl)glycidyl ether (13). Compound 12 (6.0 g, 31.9 mol) was dissolved in Et_2O (5 ml) and (*R*,*R*)-(salen)Co(III)(AcO) (15) (0.10 g, 0.15 mmol) was added. The mixture was cooled to 0°C and H₂O (0.32 ml,

17.5 mmol) was added dropwise. The reaction was allowed to reach room temperature and then stirred for 20 h. The Et₂O was evaporated and the residue was purified by bulb to bulb distillation 50–55°C/0.6 mm affording **13** (2.7 g, 45% yield). $[\alpha]_D^{25}=-2.2$ (*c*=1.52, CHCl₃) {Ref.^{13a} $[\alpha]_D^{20}=-2.20$ (*c*=0.64, CHCl₃)}. ¹H NMR (CDCl₃, 300 MHz): δ 3.8 (1H, ddd, *J*=11.8, 3.3, 0.4 Hz, ¹H of CH₂–OTES), 3.6 (1H, dd, *J*=11.8, 5.0 Hz, ¹H of CH₂–OTES), 3.1–3.0 (1H, m, CH–O), 2.8–2.7 (1H, ddd, *J*=5.1, 4.0, 0.4 Hz, ¹H of CH₂–O–CH), 2.6 (1H, dd, *J*=5.1, 2.8 Hz, ¹H of CH₂–O–CH), 0.9 (9H, t, *J*=7.8 Hz, CH₃–CH₂–Si), 0.6 (6H, q, *J*=7.8 Hz, CH₂–Si). ¹³C NMR (CDCl₃, 75.5 MHz): δ 63.6 (CH₂–OTES), 52.3 (CH–O), 44.4 (CH₂–O–CH), 6.4 (3C, CH₃–CH₂–Si), 4.2 (3C, CH₂–Si).

1.1.12. (S)-1-[(Triethylsilyl)oxy]-4-decyn-2-ol (17). The TES protected glycidol 13 (2.0 g, 10.6 mmol) was transformed in 17 following the same procedure as for the preparation of compound 16. Crude 17 (2.9 g) was obtained and used for the next step without further purification. A small sample was purified by flash chromatography for analytical purpose. $[\alpha]_D^{25}=9.2$ (c=1.75, CH₂Cl₂). ¹H NMR (CDCl₃, 300 MHz): δ 3.8–3.6 (2H, m, CH–OH, ¹H of CH₂-OTES), 3.6-3.5 (1H, m, ¹H of CH₂-OTES), 2.5 (1H, br. s, OH), 2.5-2.3 (2H, m, CH₂-CH-OH), 2.2-2.1 [2H, tt, J=7.0, 2.4 Hz, CH₂-(CH₂)₃-CH₃], 1.5-1.4 [2H, m, CH₂-(CH₂)₂-CH₃], 1.4-1.2 [4H, m, (CH₂)₂-CH₃], 1.0-0.9 (9H, t, J=8.0 Hz, CH₃-CH₂-Si), 0.9-0.8 (3H, t, J=7.2 Hz, CH₃), 0.6 (6H, q, J=8.0 Hz, CH₂-Si). ¹³C NMR (CDCl₃, 75.5 MHz): δ 82.6 (≡C), 75.6 (≡C), 70.5 (CH–OH), 65.4 (CH₂-OTES), 31.0 (CH₂-CH₂-CH₃), 28.6 [CH₂-(CH₂)₂-CH₃], 23.3 (CH₂-CH-OH), 22.1 (CH₂-CH₃), 18.6 [CH₂-(CH₂)₃-CH₃], 13.8 (CH₃), 6.5 (3C, CH₃-CH₂-Si), 4.2 (3C, CH₂–Si). Enantiomeric excess was determined to be >99%by chiral HPLC of the dibenzoate derivative obtained from compound 17 after cleavage of the TES ether [Column: OF-chiracel, mobile phase: hexane/i-PrOH (90:10), $\lambda = 265$ nm].

1.1.13. (S)-1,2-Bis[(triethylsilyl)oxy]-4-decyne (18). Following the same procedure that for the preparation of 5, 17 (2.7 g, 100 g)9.5 mmol) was transformed in 18. After drying and concentrating under vacuo the product was purified by flash chromatography with hexane/EtOAc (98:2) (0.5% Et₃N) affording **18** (2.8 g, 66% two-steps). $[\alpha]_{D}^{25} = -1.9$ (c=1.19, CH₂Cl₂). ¹H NMR (CDCl₃, 300 MHz): δ 3.8 (1H, m, CH-OTES), 3.7-3.5 (2H, 2 dd AB system, J=9.9, 5.7 Hz, CH2-OTES), 2.5-2.4 (1H, ddt AB system, J=16.5, 6.0, 2.4 Hz, ¹H of CH_2 -CH-OTES), 2.3–2.2 (1H, ddt AB system, J=16.5, 5.7, 2.4 Hz, ¹H of CH₂-CH-OTES), 2.1 [2H, tt, J=6.9, 2.4 Hz, CH₂-(CH₂)₃-CH₃], 1.5-1.4 [2H, m, $CH_2-(CH_2)_2-CH_3$], 1.4–1.2 [4H, m, (CH₂)₂-CH₃], 1.0–0.8 (21H, m, CH₃-CH₂-Si, CH₃), 0.7–0.5 (12H, m, CH₂-Si). ¹³C NMR (CDCl₃, 75.5 MHz): δ 81.6 (=C), 77.0 (=C), 72.5 (CH-OTES), 66.3 (CH₂-OTES), 31.1 (CH₂-CH₂-CH₃), 28.7 [CH₂-(CH₂)₂-CH₃], 24.6 (CH₂-CH-OTES), 22.2 (CH₂-CH₃), 18.8 [CH₂-(CH₂)₃-CH₃], 13.9 (CH₃), 6.7 (6C, CH₃-CH₂-Si), 4.9 (3C, CH₂-Si), 4.3 (3C, CH₂-Si). IR (film): 2955, 2935, 2912, 2876, 1459, 1415, 1239, 1123, 1081, 1005, 815, 743, 673 cm⁻¹. Anal. Calcd for C₂₂H₄₆O₂Si₂: C, 66.26; H, 11.63. Found: C, 66.53; H, 11.58.

1.1.14. (*S*)-2-[(**Triethylsily**]**oxy**]-4-decynal (19). Following the same procedure as described for the preparation of 4, the di-TES **18** (0.30 g, 0.75 mmol) in CH_2Cl_2 (4 ml) was converted to **19**. Crude **19** was used without further purification for the next step.

1.1.15. (S)-4-[(Triethylsilyl)oxy]-2(E)-dodecen-6-ynal (21). To the crude aldehyde 19, obtained in the previous reaction, in benzene (10 ml) was added (triphenylphosphoranylidene)acetaldehyde (8) (0.45 g, 1.5 mmol). The reaction mixture was brought to 60°C and stirred for 2 h. The solvent was evaporated under vacuo and the crude product was purified by flash chromatography [hexane/EtOAc (98:2) (1%) Et_3N] affording **21** (0.14 g, 61%). ¹H NMR (CDCl₃, 300 MHz): δ 9.6 (1H, d, J=7.8 Hz, CHO), 6.9 (1H, dd, J=15.3, 4.2 Hz, =CH-CH-OTES), 6.3 (1H, ddd, J=15.3, J=157.8, 1.5 Hz, =CH-CHO), 4.5 (1H, dddd, J=8.4, 5.7, 4.2, 1.5 Hz, CH-OTES), 2.6-2.5 (1H, ddt AB system, J=16.5, 5.7, 2.4 Hz, ¹H of CH_2 -CH-OTES), 2.4–2.3 (1H, ddt AB system, J=16.5, 8.4, 2.4 Hz, ¹H of CH₂-CH-OTES), 2.1 [2H, tt, J=7.2, 2.4 Hz, CH₂-(CH₂)₃-CH₃], 1.5-1.4 [2H, m, CH₂-(CH₂)₂-CH₃], 1.4-1.2 [4H, m, (CH₂)₂-CH₃], 1.0-0.8 (12H, m, CH₃, CH₃-CH₂-Si), 0.6 (6H, m, CH₂-Si). ¹³C NMR (CDCl₃, 75.5 MHz): δ 193.3 (CHO), 158.0 (=CH-CH–OTES), 131.2 (=CH–CHO), 83.5 (≡C), 75.2 (≡C), 70.9 (CH-OTES), 31.0 (CH2-CH2-CH3), 28.5 [CH2-(CH₂)₂-CH₃], 28.1 (CH₂-CH-OTES), 22.1 (CH₂-CH₃), 18.7 [CH₂-(CH₂)₃-CH₃], 13.8 (CH₃), 6.6 (3C, CH₃-CH₂-Si), 4.8 (3C, CH₂-Si). IR (film): 2956, 2934, 2914, 2876, 2808, 2719, 1696, 1459, 1239, 1107, 1007, 975, 745, 730 cm⁻¹. Anal. Calcd for $C_{18}H_{32}O_2Si_2 \cdot 0.^2H_2O$: C, 69.26; H, 10.46. Found: C, 68.83; H, 9.96.

1.1.16. (S)-12-[(Triethylsilyl)oxy]-5(Z),8(Z),10(E)-eicosatrien-14-ynoic acid methyl ester (24). Wittig reaction of aldehyde 21 (0.10 g, 0.32 mmol) with the phosphorane 9, obtained in situ from the phosphonium iodide salt 22 (0.30 g, 0.56 mmol), under the same conditions as described for the preparation of compound 23, afforded 24 (0.13 g, 87%) after flash chromatography purification. $[\alpha]_D^{25}=21.3$ (c=1, acetone). ¹H NMR (d₆-benzene, 300 MHz): δ 6.9 (1H, ddt, J=15.2, 11.1, 1.2 Hz, CH=CH-CH-OTES), 6.2-6.1 (1H, br. t, J=11.1 Hz, =CH-CH=CH-CH-OTES), 6.0 (1H, dd, J=15.2, 6.0 Hz, =CH-CH-OTES), 5.6–5.3 [3H, m, =CH–CH₂–CH=CH–(CH₂)₃–COO], 4.5 (1H, m, CH-OTES), 3.4 (3H, s, CH₃O), 3.0 [2H, br. t, J=7.4 Hz, CH₂-CH=CH-(CH₂)₃-COO], 2.7-2.6 (1H, ddt AB system, J=16.2, 6.0, 2.4 Hz, ¹H of CH₂-CH-OTES), 2.6–2.5 (1H, ddt AB system, J=16.2, 7.0, 2.4 Hz, ¹H of CH₂-CH-OTES), 2.2 [4H, m, CH₂-(CH₂)₃-CH₃, CH2-COO], 2.0 [2H, m, CH2-(CH2)2-COO], 1.8-1.6 (2H, quint, J=7.2 Hz, CH₂-CH₂-COO), 1.6-1.5 [2H, m, CH₂-(CH₂)-COO], 1.5-1.2 [4H, m, (CH₂)₂-CH₃], 1.1 (9H, t, J=8.0 Hz, CH₃-CH₂-Si), 0.9 (3H, t, J=7.9 Hz, CH₃), 0.7 (6H, q, J=7.9 Hz, CH₂-Si). ¹³C NMR (CDCl₃, 75.5 MHz): δ 174.1 (COO), 135.9 (=CH-CH-OTES), 129.8 (=CH), 129.3 [=CH-(CH₂)₃-COO], 128.6 (=CH), 128.2 (=CH-CH=CH-CH-OTES), 124.8 (CH=CH-CH-OTES), 82.2 (≡C), 76.6 (≡C), 72.1 (CH–OTES), 51.3 (CH₃–O), 33.3 (CH₂-COO), 31.0 (CH₂-CH₂-CH₃), 28.9 (CH₂-CH-OTES), 28.6 $[CH_2-(CH_2)_2-CH_3]$, 26.4 $[CH_2-(CH_2)_2-CH_3]$ COO], 25.9 (=CH-CH₂-CH=), 24.7 (CH₂-CH₂-COO), 22.1 (CH₂-CH₃), 18.7 [CH₂-(CH₂)₃-CH₃], 13.8 (CH₃), 6.7

(3C, CH_3-CH_2-Si), 4.8 (3C, CH_2-Si). IR (film): 3011, 2953, 2933, 2875, 1741, 1457, 1435, 1239, 1155, 1107, 1072, 1005, 984, 951, 743, 727 cm⁻¹. Anal. Calcd for $C_{27}H_{46}O_3Si$: C, 72.59; H, 10.38. Found: C, 72.30; H, 10.76.

1.1.17. (S)-12-[(Triethylsilyl)oxy]-5(Z),8(Z),10(E),14(Z)eicosatetraenoic acid methyl ester (26). The alkyne 24 (67.0 mg, 0.15 mmol) was dissolved in hexane (4 ml) under argon and pyridine (70 µL) followed by Lindlar catalyst (100 mg) were added. The argon atmosphere was exchanged by hydrogen and the mixture was stirred for 36 h. Filtration of the catalyst, evaporation of the solvent and purification by flash chromatography with hexane/EtOAc (95:5) (1% Et₃N) afforded **26** (61 mg, 90%). ¹H NMR (CDCl₃, 300 MHz): δ 6.4 (1H, ddt, *J*=15.3, 11.1, 1.2 Hz, CH=CH-CH-OTES), 6.0 (1H, br. t, J=11.1 Hz, =CH-CH=CH-CH-OTES), 5.7-5.6 (1H, dd, J=15.3, 6.3 Hz, =CH-CH-OTES), 5.5-5.2 [5H, m, =CH-CH₂- $CH = CH - (CH_2)_3 - COO, CH = CH - (CH_2)_4 - CH_3],$ (1H, m, CH-OTES), 3.6 (3H, s, CH₃O), 2.9 (2H, m, =CH-CH $_2-$ CH=), 2.3 (2H, t, J=7.5 Hz, CH $_2-$ COO), 2.3-2.2 (2H, m, CH2-CH-OTES), 2.1 [2H, m, CH2-(CH₂)₂-COO], 2.0 [2H, m, CH₂-(CH₂)₃-CH₃], 1.7 (2H, quint, J=7.5 Hz, CH_2-CH_2-COO), 1.4–1.2 [6H, m, (CH₂)₃-CH₃], 1.0-0.9 (9H, m, CH₃-CH₂-Si), 0.9 (3H, t, J=6.8 Hz, CH₃), 0.6 (6H, m, CH₂-Si). ¹³C NMR (CDCl₃, 75.5 MHz): δ 174.1 (COO), 137.0 (=CH-CH-OTES), 132.1 [=*C*H-(CH₂)₄-CH₃], 129.5 (*C*H=CH-CH=CH- $[=CH-(CH_2)_3-COO],$ 129.3 CH-OTES), 128.7 [CH=CH-(CH₂)₃-COO], 128.3 (=CH-CH=CH-CH-OTES), 125.2 [CH=CH-(CH₂)₄-CH₃], 124.4 (CH=CH-CH-OTES), 73.0 (CH-OTES), 51.4 (CH₃-O), 36.4 (CH₂-CH-OTES), 33.4 (CH₂-COO), 31.5 (CH₂-CH₂-CH₃), 29.2 [CH₂-(CH₂)₂-CH₃], 27.4 [CH₂-(CH₂)₃-CH₃], 26.5 $[CH_2-(CH_2)_2-COO], 26.0 (=CH-CH_2-CH=), 24.7$ (CH₂-CH₂-COO), 22.5 (CH₂-CH₃), 13.9 (CH₃), 6.8 (3C, CH₃-CH₂-Si), 4.9 (3C, CH₂-Si). IR (film): 3011, 2953, 2931, 2875, 1743, 1457, 1239, 1073, 1005, 743, 727 cm⁻¹.

1.1.18. 12(S)-HETE methyl ester (28). To compound 26 (66 mg, 0.15 mmol) in MeOH (7 ml) was added pyridinium p-toluenesulfonate (10 mg). The mixture was stirred for 1.5 h at room temperature and then quenched by the addition of Et₃N. The solvent was evaporated under vacuo and the crude was purified by flash chromatography with hexane/EtOAc (9:1) affording 12(S)-HETE methyl ester (45 mg, 90%). $[\alpha]_D^{25}=11.4$ (c=1.1, acetone). {Ref.^{10f} $[\alpha]_D^{20} = 13$ (c=1.5, acetone)}. UV (CH₃CN) λ_{max} 237 nm. ¹H NMR (CDCl₃, 300 MHz): δ 6.5 (1H, ddt, J=15.0, 11.1, 1.2 Hz, CH=CH-CH-OH), 6.0-5.9 (1H, br. t, J=11.1 Hz, =CH-CH=CH-CH-OH), 5.7 (1H, dd, J=15.0, 6.0 Hz, =CH-CH-OH), 5.6-5.5 [1H, m, =CH- $(CH_2)_4-CH_3$], 5.4–5.3 [4H, m, =CH-CH_2-CH=CH-(CH₂)₃-COO, CH=CH-(CH₂)₄-CH₃], 4.2 (1H, m, CH-OH), 3.6 (3H, s, CH₃O), 2.9 (2H, m, =CH-CH₂-CH=), 2.4-2.2 (4H, t and m, J=7.5 Hz, CH₂-COO, CH₂-CH-OH), 2.2–2.0 [4H, m, CH₂–(CH₂)₂–COO, CH₂–(CH₂)₃– CH₃], 2.0–1.9 (1H, br. s, OH), 1.8–1.6 (2H, m, CH₂– CH_2 -COO), 1.4–1.2 [6H, m, (CH_2)₃– CH_3], 0.9 (3H, t, J=6.8 Hz, CH₃). ¹³C NMR (CDCl₃, 75.5 MHz): δ 174.1 (COO), 136.0 (=*C*H-CH-OH), 133.6 [=*C*H-(CH₂)₄-CH₃], 130.2, 129.4, 128.5 [=CH-CH₂-CH=CH-(CH₂)₃-COO], 128.1 (=CH-CH=CH-CH-OH), 125.4 1.1.19. 12(S)-HETE (2). Compound 28 (15.0 mg, 0.045 mmol) was converted to 2 using the same conditions as for the preparation of 1 affording 12(S)-HETE (13 mg, 90%). UV (CH₃CN) λ_{max} 237 nm. ¹H NMR (CDCl₃, 300 MHz): δ 6.6 (1H, br. dd, J=15.3, 11.0 Hz, CH=CH-CH-OH), 6.0 (1H, br. t, J=11.0 Hz, =CH-CH=CH-CH-OH), 5.7 (1H, dd, J=15.3, 6.3 Hz, =CH-CH-OH), 5.6–5.5 [1H, dtt, J=11.1, 7.2, 1.5 Hz, $=CH-(CH_2)_4-$ CH₃], 5.5–5.3 [4H, m, =CH–CH₂–CH=CH–(CH₂)₃– COO, CH=CH-(CH₂)₄-CH₃], 4.3-4.2 (1H, m, CH-OH), 3.0-2.8 (2H, m, =CH-CH₂-CH=), 2.4-2.3 (4H, m, CH₂-COO, CH_2 -CH-OH), 2.1 [2H, m, CH_2 -(CH₂)₂-COO], 2.1-2.0 [2H, m, CH₂-(CH₂)₃-CH₃], 1.7 (2H, m, CH₂-CH2-COO), 1.4-1.2 [6H, m, (CH2)3-CH3], 0.9-0.8 (3H, t, J=6.8 Hz, CH₃). ¹³C NMR (CDCl₃, 75.5 MHz): δ 178.2 (COO), 135.4 (=CH-CH-OH), 133.7 [=CH-(CH₂)₄-CH₃], 130.3, 129.4, 128.4 [=*C*H-CH₂-*C*H=*C*H-(CH₂)₃-COO], 127.8 (=CH-CH=CH-CH-OH), 125.6 $[CH = CH - (CH_2)_4 - CH_3],$ (CH = CH - CH - OH),124.3 72.1 (CH-OH), 35.3 (CH2-CH-OH), 33.5 (CH2-COO), 31.4 (CH₂-CH₂-CH₃), 29.2 [CH₂-(CH₂)₂-CH₃], 27.4 $[CH_2-(CH_2)_3-CH_3], 26.4 [CH_2-(CH_2)_2-COO],$ 26.2 $(=CH-CH_2-CH=), 24.6$ $(CH_2 - CH_2 - COO),$ 22.4 (CH₂-CH₃), 13.9 (CH₃). HPLC/API-ES/MS (m/z): 319 $[M_{12(S)-HETE} - H^+]^-$.

1.1.20. (S)-1,2-Bis[(triethylsilyl)oxy]-4(Z)-decene (30). To a solution of alkyne 18 (0.30 g, 0.75 mmol) in hexane (10 ml) under argon was added Lindlar catalyst (50 mg) followed by Et_3N (50 µL). The atmosphere of argon was exchanged by hydrogen and the mixture was vigorously stirred for 1 h. The catalyst was filtered and the solvent evaporated under vacuo. The crude product 30 was used directly in the next step. $[\alpha]_D^{25} = -2.3$ (c=1 crude product, CH₂Cl₂). ¹H NMR (CDCl₃, 300 MHz): δ 5.5–5.3 (2H, m, CH=CH), 3.7 (1H, m, CH-OTES), 3.6-3.4 (2H, m, CH₂-OTES), 2.4–2.3 (1H, br. dt, J=14.4, 5.7 Hz, ¹H of CH_2 – CH-OTES), 2.3–2.1 (1H, br. dt, J=14.4, 6.0 Hz, ¹H of CH_2 -CH-OTES), 2.0 [2H, m, CH_2 -(CH₂)₃-CH₃], 1.4-1.2 [6H, m, (CH₂)₃-CH₃], 1.0-0.8 (21H, br. t and t, J=7.8 Hz, J=6.9 Hz, CH₃-CH₂-Si, CH₃), 0.7–0.5 (12H, 2q, J=7.8 Hz, CH₂-Si). ¹³C NMR (CDCl₃, 75.5 MHz): δ 131.9 (=CH), 125.6 (=CH), 73.4 (CH-OTES), 66.8 (CH2-OTES), 32.3 (CH2-CH-OTES), 31.5 (CH2-CH2-CH₃), 29.3 [CH₂-(CH₂)₂-CH₃], 27.3 [CH₂-(CH₂)₃-CH₃], 22.5 (CH₂-CH₃), 13.9 (CH₃), 6.7 (3C, CH₃-CH₂-Si), 6.6 (3C, CH₃-CH₂-Si), 4.9 (3C, CH₂-Si), 4.3 (3C, CH₂-Si).

1.1.21. (*S*)-2-[(**Triethylsily**])**oxy**]-4(*Z*)-**decenal** (32). The di-TES **30** (0.20 g, 0.50 mmol) was oxidized following the same Swern conditions as described for the preparation of **4**. The crude product **32** was used without further purification for the next step.

1.1.22. (S)-4-[(Triethylsilyl)oxy]-2(E),6(Z)-dodecadienal (34). Wittig reaction of 32, obtained in the previous

reaction, with 8 afforded aldehyde 34 using the conditions described for the preparation of **20**. The crude was purified by flash chromatography [hexane/EtOAc (98:2) (1% Et₃N)] affording **34** (91 mg, 59% two-steps). $[\alpha]_{D}^{25}=21.7$ (c=0.94, CH₂Cl₂). ¹H NMR (CDCl₃, 300 MHz): δ 9.5 (1H, d, J=7.8 Hz, CHO), 6.8 (1H, dd, J=15.5, 4.5 Hz, =CH-CH-OTES), 6.4-6.2 (1H, ddd, J=15.5, 7.8, 1.6 Hz, =CH-CHO), 5.5 [1H, m, =CH-(CH₂)₄-CH₃], 5.3 [1H, m, CH=CH-(CH₂)₄-CH₃], 4.4 (1H, m, CH-OTES), 2.4-2.2 (2H, m, CH₂-CH-OTES), 2.0 [2H, m, CH₂-(CH₂)₃-CH₃], 1.4–1.2 [6H, m, (CH₂)₃–CH₃], 1.0–0.9 (9H, t, J=7.8 Hz, CH_3-CH_2-Si), 0.9–0.8 (3H, t, J=6.9 Hz, CH₃), 0.8 (6H, q, J=7.8 Hz, CH₂–Si). ¹³C NMR (CDCl₃, 75.5 MHz): δ 193.6 (CHO), 159.5 (=CH-CH-OTES), 133.4 [=CH-(CH_2)₄- CH_3], 130.8 (=CH-CHO), 123.6 [CH=CH-(CH₂)₄-CH₃], 71.5 (CH-OTES), 35.4 (CH₂-CH-OTES), 31.4 (CH₂-CH₂-CH₃), 29.1 [CH₂-(CH₂)₂- CH_3], 27.4 [CH_2 -(CH_2)₃- CH_3], 22.4 (CH_2 - CH_3), 13.9 (CH₃), 6.6 (3C, CH₃-CH₂-Si), 4.7 (3C, CH₂-Si). IR (film): 3011, 2956, 2929, 2876, 2858, 2808, 2719, 1695, 1459, 1414, 1239, 1141, 1102, 1006, 955, 743, 729 cm⁻¹. Anal. Calcd for C₁₈H₃₄O₂Si: C, 69.62; H, 11.03. Found: C, 68.95; H, 11.63.

1.1.23. (*S*)-12-[(Triethylsilyl)oxy]-5(*Z*),8(*Z*),10(*E*),14(*Z*)eicosatetraenoic acid methyl ester (26). Wittig reaction of aldehyde 34 (71.4 mg, 0.23 mmol) with the phosphorane 9, obtained in situ from the phosphonium iodide salt 22 (0.20 g, 0.40 mmol), under the same conditions as described for the preparation of compound 23, afforded 26 (85 mg, 82%) after flash chromatography. $[\alpha]_D^{25}=13.4$ (*c*=1.1, acetone).

1.1.24. 4,5-Dideuterio-1,2(R)-bis[(triethylsilyl)oxy]-4(Z)decene (29). Following the same conditions used for the transformation of 18 to 30, alkyne 5 (0.30 g, 0.75 mmol) was reduced with deuterium in the presence of Lindlar catalyst (100 mg) and Et₃N (80 µl). After 30 min the starting material was consumed, the reaction mixture was filtered and the solvent was evaporated affording 29 that was used without further purification in the next step. $[\alpha]_D^{25} = -2.12$ (c=2.1, CH₂Cl₂). ¹H NMR (CDCl₃, 300 MHz): § 3.7 (1H, m, CH-OTES), 3.6-3.4 (2H, 2 dd AB system, J=9.9, 5.7 Hz, CH₂-OTES), 2.4-2.3 (1H, dd AB system, J=14.4, 5.7 Hz, ¹H of CH₂-CH-OTES), 2.2-2.1 (1H, dd AB system, J=14.4, 6.0 Hz, ¹H of CH_2 -CH-OTES), 2.0 [2H, br. t, J=6.9 Hz, CH₂-(CH₂)₃-CH₃], 1.4-1.2 [6H, m, (CH₂)₃-CH₃], 1.0-0.9 [21H, br. t, J=7.8 Hz, t, J=6.9 Hz, $(CH_3-CH_2)_3-Si$, CH_3], 0.7–0.5 (12H, 2q, J=7.8 Hz, CH_2-Si). ¹³C NMR (CDCl₃, 75.5 MHz): δ 131.4 (t, J=23 Hz, =CD), 125.2 (t, J=23 Hz, =CD), 73.4 (CH-OTES), 66.8 (CH₂-OTES), 32.2 (CH₂-CH-OTES), 31.5 (CH₂-CH₂-CH₃), 29.3 [CH₂-(CH₂)₂-CH₃], 27.2 $[CH_2-(CH_2)_3-CH_3]$, 22.5 (CH_2-CH_3) , 13.9 (CH_3) , 6.7 (3C, CH₃-CH₂-Si), 6.6 (3C, CH₃-CH₂-Si), 4.9 (3C, CH₂-Si), 4.3 (3C, CH₂-Si). IR (film): 2955, 2934, 2913, 2876, 2248, 1630, 1459, 1415, 1239, 1121, 1087, 1005, 811, 742, 728 cm⁻¹. Anal. Calcd for C₂₂H₄₆D₂O₂Si₂: C, 65.60; H, 12.51. Found: C, 64.83; H, 12.18.

1.1.25. 4,5-Dideuterio-2(R)-[(triethylsilyl)oxy]-4(Z)-decenal (31). The di-TES 29 (0.20 g, 0.50 mmol) was oxidized following the same Swern conditions as described for the

preparation of **4**. The crude product **31** was used without further purification in the next step. ¹H NMR (CDCl₃, 300 MHz): δ 9.6 (1H, d, *J*=1.8 Hz, CHO), 3.9 (1H, td, *J*=6.3, 1.8 Hz, CH–OTES), 2.4–2.3 (2H, d, *J*=6.3 Hz, CH₂–CH–OTES), 2.0–1.9 [2H, t, *J*=7.4 Hz, CH₂–(CH₂)₃–CH₃], 1.4–1.2 [6H, m, (CH₂)₃–CH₃], 1.0–0.8 (12H, m, CH₃, CH₃–CH₂–Si), 0.6–0.4 (6H, m, CH₂–Si). ¹³C NMR (CDCl₃, 75.5 MHz): δ 203.6 (CHO), 132.9 (t, *J*=23 Hz, =CD), 122.6 (t, *J*=23 Hz, =CD), 77.4 (CH–OTES), 31.4 (CH₂–CH₂–CH₃), 30.9 (CH₂–CH–OTES), 29.0 [CH₂–(CH₂)₂–CH₃], 27.1 [CH₂–(CH₂)₃–CH₃], 22.4 (CH₂–CH₃), 13.8 (CH₃), 6.4 (3C, CH₃–CH₂–Si).

1.1.26. 6,7-Dideuterio-4(R)-[(triethylsilyl)oxy]-2(E),6(Z)dodecadienal (33). Wittig reaction of 31, obtained in the previous reaction, with 8 afforded aldehyde 33 using the conditions described for the preparation of 20. The crude was purified by flash chromatography [SiO₂, hexane/EtOAc (98:2) (1% Et₃N)] affording **33** (90 mg, 59% two-steps). ¹H NMR (CDCl₃, 300 MHz): δ 9.6 (1H, d, *J*=7.8 Hz, CHO), 6.8 (1H, dd, J=15.6, 4.5 Hz, =CH-CH-OTES), 6.3 (1H, ddd, J=15.6, 7.8, 1.5 Hz, =CH-CHO), 4.4 (1H, dddd, J=6.9, 6.0, 4.5, 1.5 Hz, CH-OTES), 2.4 (2H, m, CH₂-CH-OTES), 2.0 [2H, m, CH₂-(CH₂)₃-CH₃], 1.4-1.2 [6H, m, (CH₂)₃-CH₃], 1.0-0.9 (9H, t, J=7.9 Hz, CH₃-CH₂-Si), 0.9-0.8 (3H, t, J=6.8 Hz, CH₃), 0.6 (6H, q, J=7.9 Hz, CH_2 -Si). ¹³C NMR (CDCl₃, 75.5 MHz): δ 193.5 (CHO), 159.3 (=CH-CH-OTES), 133.0 [t, J=23 Hz, =CD-(CH₂)₄-CH₃], 131.0 (=CH-CHO), 123.2 [t, J=23 Hz, CD=CD-(CH₂)₄-CH₃], 71.7 (CH-OTES), 35.4 (CH₂-CH-OTES), 31.5 (CH₂-CH₂-CH₃), 29.1 [CH₂-(CH₂)₂-CH₃], 27.3 [CH₂-(CH₂)₃-CH₃], 22.4 (CH₂-CH₃), 13.8 (CH₃), 6.6 (3C, CH₃-CH₂-Si), 4.8 (3C, CH₂-Si). IR (film): 2956, 2928, 2876, 2856, 2807, 2718, 2249, 1695, 1459, 1239, 1139, 1104, 1005, 976, 744, 729 cm⁻¹. Anal. Calcd for C₁₈H₃₂D₂O₂Si: C, 69.17; H, 11.61. Found: C, 69.11; H, 11.59.

1.1.27. 14,15-Dideuterio-12(R)-[(triethylsilyl)oxy]-5(Z),8(Z), 10(E), 14(Z)-eicosatetraenoic acid methyl ester (35). Wittig reaction of aldehyde 33 (0.10 g, 0.32 mmol) with the phosphorane 9, obtained in situ from the phosphonium iodide salt 22 (0.27 g, 0.5 mmol), under the same conditions as described for the preparation of compound 23, afforded **35** (0.11 g, 84%) after flash chromatography purification. ¹H NMR (d₆-benzene, 300 MHz): δ 6.8 (1H, br. dd, J=15.3, 11.1 Hz, CH=CH-CH-OTES), 6.1 (1H, br. t, J=11.1 Hz, =CH-CH=CH-CH-OTES), 5.8 (1H, dd, J=15.3, 6.3 Hz, =CH-CH-OTES), 5.5-5.2 [3H, m, =CH-CH₂-CH=CH-(CH₂)₃-COO], 4.4-4.3 (1H, m, CH-OTES), 3.4 (3H, s, CH₃O), 3.0–2.9 (2H, m, =CH-CH₂-CH=), 2.6-2.4 (2H, 2 dd AB system, J=14.1, 6.6 Hz, CH₂-CH-OTES), 2.2–2.1 (2H, t, J=7.5 Hz, CH₂–COO), 2.1 [2H, m, $CH_2-(CH_2)_3-CH_3$, 2.1–1.9 [2H, m, $CH_2-(CH_2)_2-COO$], 1.7-1.6 (2H, m, CH₂-CH₂-COO), 1.5-1.2 [6H, m, (CH₂)₃-CH₃], 1.2-1.0 (9H, t, J=7.8 Hz, CH₃-CH₂-Si), 0.9 (3H, t, J=6.8 Hz, CH₃), 0.8–0.6 (6H, q, J=7.8 Hz, CH₂-Si). ¹³C NMR (d₆-benzene, 75.5 MHz): δ 173.1 (COO), 137.5 (=CH-CH-OTES), 129.8, 129.6, 128.7 $[=CH-CH_2-CH=CH-(CH_2)_3-COO], 127.9 (=CH-CH_2)_3-COO], 127.9 (=CH-CH$ CH=CH-CH-OTES), 124.9 (CH=CH-CH-OTES), 73.5 (CH-OTES), 50.7 (CH₃-O), 36.8 (CH₂-CH-OTES), 33.3 (CH2-COO), 31.7 (CH2-CH2-CH3), 29.5 [CH2 $(CH_2)_2-CH_3]$, 27.6 $[CH_2-(CH_2)_3-CH_3]$, 26.7 $[CH_2-(CH_2)_2-COO]$, 26.3 (=CH- $CH_2-CH=$), 24.9 (CH_2-CH_2-COO) , 22.7 (CH_2-CH_3) , 14.0 (CH_3) , 6.9 (3C, CH_3-CH_2-Si), 5.4 (3C, CH_2-Si). The ¹³C signals corresponding to the carbons CD=CD were not observed due to their low intensity. IR (film): 3011, 2953, 2929, 2875, 2247, 1742, 1456, 1435, 1239, 1071, 1005, 983, 950, 742, 726 cm⁻¹. HRMS (ESI) calcd for $C_{27}H_{46}D_2O_3SiNa$ $(M+Na)^+$: 473.3396, found $(M+Na)^+$: 473.3408.

1.1.28. 14,15-Dideuterio-12(R)-HETE methyl ester (36). To compound 35 (50 mg, 0.11 mmol) in MeOH (5 ml) was added pyridinium p-toluenesulfonate (10 mg). The mixture was stirred for 1.5 h at room temperature and then quenched by the addition of Et₃N. The solvent was evaporated under vacuo and the crude was purified by flash chromatography with hexane/EtOAc (9:1) affording 36 (34 mg, 92%). UV (CH₃CN) λ_{max} 237 nm. ¹H NMR (CDCl₃, 300 MHz): δ 6.5 (1H, ddt, J=15.3, 11.1, 1.2 Hz, CH=CH-CH-OH), 6.0 (1H, br. t, J=11.1 Hz, =CH-CH=CH-CH-OH), 5.7 (1H, dd, J=15.3, 6.3 Hz, =CH-CH-OH), 5.4-5.3 [3H, m, $=CH-CH_2-CH=CH-(CH_2)_3-COO]$, 4.2 (1H, m, CH-OH), 3.6 (3H, s, CH₃O), 2.9 (2H, m, =CH-CH₂-CH=), 2.4-2.2 (4H, m, CH₂-COO, CH₂-CH-OH), 2.2-2.0 [4H, m, CH₂-(CH₂)₂-COO, CH₂-(CH₂)₃-CH₃], 1.8–1.6 (2H, m, CH_2 – CH_2 –COO), 1.4–1.2 [6H, m, $(CH_2)_3$ – CH_3], 0.9 (3H, t, J=6.8 Hz, CH_3). ¹³C NMR (CDCl₃, 75.5 MHz): δ 174.2 (COO), 136.0 (=CH-CH-OH), 133.2 [t, J=23 Hz, =CD-(CH₂)₄-CH₃], 130.3, 128.4 $=CH-CH_2-CH=CH-(CH_2)_3-COO],$ 129.3. 128.0 (=CH-CH=CH-CH-OH), 125.4 (CH=CH-CH-OH), 123.9 [t, J=23 Hz, CD=CD-(CH₂)₄-CH₃], 72.0 (CH-OH), 51.4 (CH₃-O), 35.2 (CH₂-CH-OH), 33.4 (CH₂-COO), 31.4 (CH₂-CH₂-CH₃), 29.2 [CH₂-(CH₂)₂-CH₃], 27.2 [CH₂-(CH₂)₃-CH₃], 26.5 [CH₂-(CH₂)₂-COO], 26.0 (=CH-CH₂-CH=), 24.7 (CH₂-CH₂-COO), 22.5 (CH₂-CH₃), 13.9 (CH₃). IR (film): 3457 (br), 3009, 2953, 2927, 2856, 2248, 1739, 1437, 1156, 984, 950 cm^{-1} . HRMS (ESI) calcd for C₂₁H₃₂D₂O₃Na $(M+Na)^+$: 359.2531, found $(M+Na)^+$: 359.2539.

1.1.29. [14,15-²H₂]-12(*R*)-HETE (37). Compound 36 (1.6 mg, 4.8 µmol) was converted to 37 using the same conditions as for the preparation of 1 affording $[14,15-{}^{2}H_{2}]-12(R)$ -HETE (1.4 mg, 91%). UV (CH₃CN) λ_{max} 237 nm. ¹H NMR (CD₃CN, 300 MHz): δ 6.6–6.5 (1H, br. dd, J=15.3, 11.1 Hz, CH=CH-CH-OH), 6.0 (1H, br. t, J=11.1 Hz, =CH-CH=CH-CH-OH), 5.7 (1H, dd, J=15.3, 6.3 Hz, =CH-CH-OH), 5.5-5.3 [3H, m, =CH-CH₂-CH=CH-(CH₂)₃-COO], 4.2-4.1 (1H, m, CH-OH), 2.9 (2H, m, =CH-CH₂-CH=), 2.4-2.0 (8H, m, CH₂-CH-OH, CH₂-CH₂-CH₂-COO, CH₂-(CH₂)₃-CH₃], 1.7–1.5 (2H, m, CH₂–CH₂–COO), 1.4–1.2 [6H, m, $(CH_2)_3$ -CH₃, 0.9 (3H, t, J=6.8 Hz, CH₃). IR (film): 3369 (br), 3009, 2955, 2927, 2857, 2248, 1709, 1456, 1365, 1245, 1159, 1048, 983, 950 cm⁻¹. HRMS (ESI) calcd for $C_{20}H_{30}D_2O_3Na$ (M+Na)⁺: 345.2375, found (M+Na)⁺: 345.2390.

1.1.30. 5(S)-[(*tert*-Butyldiphenylsilyl)oxy]-12(*R*)-[(triethyl-silyl)oxy]-8(*E*),10(*E*)-eicosadien-6,14-diynoic acid methyl ester (38). Phosphonate 10 (0.40 g, 0.74 mmol) was dissolved in THF (7 ml) and cooled to -78° C. A 1.5 M

solution of LDA in cyclohexane (0.53 ml, 0.79 mmol) was added dropwise. After 2 min aldehyde **4** (0.19 g, 0.67 mmol) in THF (2 ml) was added. The reaction was stirred for 1 h at -78° C and then was allowed to reach room temperature before quenching with NH₄Cl aq/Et₂O. The organic phase was separated, washed with brine, dried (Na₂SO₄) and concentrated under vacuo. The crude product was filtered through silica affording a mixture of *cis* and *trans* isomers at the C₁₀-C₁₁ double bond.

The crude mixture was dissolved in CH₂Cl₂ (10 ml) and treated with a catalytic amount of I₂. After stirring at room temperature for 30 min the reaction was quenched by addition of a concentrated solution of Na₂S₂O₃ (1 ml). Separation of phases, drying over Na₂SO₄, concentrating and flash chromatography [hexane/EtOAc (98:2)] afforded **38** (0.23 g, 49%). ¹H NMR (CDCl₃, 300 MHz): δ 7.8–7.6 (4H, m, Ar), 7.5–7.3 (6H, m, Ar), 6.4–6.1 (2H, m, =CH– CH=CH-CH-OTES), 5.8 (1H, dd, J=13.8, 5.7 Hz, =CH-CH-OTES), 5.4 (1H, m, =CH-C \equiv C-CH-OTBDPS), 4.5-4.4 (1H, m, CH-OTBDPS), 4.3-4.2 (1H, m, CH-OTES), 3.6 (3H, s, CH₃O), 2.5-2.2 (4H, m, CH₂-COO, CH₂-CH-OTES), 2.2-2.1 [2H, m, CH₂-(CH₂)₃-CH₃], 1.8–1.6 [4H, m, (CH₂)₂–CH₂–COO], 1.5–1.4 [2H, m, CH_2 -(CH_2)₂- CH_3], 1.4-1.2 [4H, m, (CH_2)₂- CH_3], 1.1 [9H, s, (CH₃)₃C], 1.0–0.9 [9H, t, J=7.8 Hz, (CH₃–CH₂)₃Si], 0.9 (3H, t, J=6.9 Hz, CH₃), 0.6 [6H, q, J=7.8 Hz, (CH₃-CH₂)₃Si]. ¹³C NMR (CDCl₃, 75.5 MHz): δ 174.0 (COO), 141.1 (CH=CH-C=), 138.1 (=CH-CH-OTES), 136.2, 136.0 (4C, ArCH), 133.9, 133.7 (2C, ArC), 129.8, 129.6 (2C, ArCH), 129.1 (CH=CH-CH-OTES), 127.7, 127.4 (4C, ArCH), 110.7 (=CH-C=), 92.7 (C=), 84.6 (C=),82.4 (C=), 76.6 (C=), 71.9 (CH-OTES), 64.0 (CH-OTBDPS), 51.4 (CH₃O), 37.6 [CH₂-(CH₂)₂-COO], 33.7 (CH₂-COO), 31.0 (CH₂-CH₂-CH₃), 28.8, 28.6 [CH₂-CH-OTES, CH₂-(CH₂)₂-CH₃], 26.9 [(CH₃)₃C], 22.1 (CH₂-CH₃), 20.4 (CH₂-CH₂-COO), 19.2 (C), 18.7 [CH₂-(CH₂)₃-CH₃], 13.9 (CH₃), 6.7 [(CH₃-CH₂)₃Si], 4.8 [(CH₃-CH₂)₃Si]. IR (film): 3070, 3047, 2954, 2931, 2874, 2858, 1741, 1460, 1428, 1361, 1241, 1171, 1149, 1111, 1080, 1005, 985, 822, 740, 701 cm⁻¹. HRMS (ESI) calcd for $C_{43}H_{62}O_4Si_2Na (M+Na)^+$: 721.4084, found $(M+Na)^+$: 721.4115.

1.1.31. 5(S)-[(tert-Butyldiphenylsilyl)oxy]-12(R)-[(triethylsilyl)oxy]-6(Z),8(E),10(E),14(Z)-eicosatetraenoicacid methyl ester (39). Compound 38 (30 mg, 0.043 mmol) was dissolved in hexane (10 ml) under an argon atmosphere and Lindlar catalyst (50 mg) was added. The argon atmosphere was exchanged by hydrogen and the mixture was stirred until TLC check showed completion. Dilution of the crude reaction with Et₂O, filtration of the catalyst, concentration and purification by flash chromatography [hexane/EtOAc (96:4)] afforded **39** (27 mg, 89%). ¹H NMR (d_6 -benzene, 300 MHz): δ 7.8–7.7 (4H, m, Ar), 7.3-7.2 (6H, m, Ar), 6.1-6.0 (1H, m, CH=CH-CH-OTES), 5.8 (2H, m, CH=CH-CH=CH-CH-OTES), 5.8–5.7 (1H, m, CH=CH-CH-OTBDPS), 5.7–5.6 (1H, dd, J=14.9, 6.8 Hz, =CH-CH-OTES), 5.5 [2H, m, $CH = CH - (CH_2)_4 - CH_3$, 5.5–5.4 (1H, dd, J = 10.5, 9.3 Hz, =CH-CH-OTBDPS), 4.7-4.6 (1H, m, CH-OTBDPS), 4.2-4.1 (1H, m, CH-OTES), 3.2 (3H, s, CH₃O), 2.5-2.2 (2H, m, CH₂-CH-OTES), 2.1-1.9 [4H, m, CH₂-COO, $CH_2-(CH_2)_3-CH_3$, 1.7–1.4 [4H, m, (CH₂)₂–CH₂–COO], 1.4–1.2 [6H, m, $(CH_2)_3$ –CH₃], 1.2 [9H, s, $(CH_3)_3$ C], 1.0 [9H, t, J=7.8 Hz, $(CH_3-CH_2)_3$ Si], 0.9-0.8 (3H, t, J=6.9 Hz, CH₃), 0.6 [6H, q, J=7.8 Hz, (CH₃-CH₂)₃Si]. ¹³C NMR (d₆-benzene, 75.5 MHz): δ 173.0 (COO), 137.8 (=CH-CH-OTES), 136.5, 136.4 (4C, ArCH), 134.7, 134.5 (2C, ArC), 134.2 (=CH-CH-OTBDPS), 133.8 (CH=), 132.2 [=*C*H-(CH₂)₄-CH₃], 129.9, 129.8 (2C, ArCH), 129.9 (CH=CH-CH-OTES), 128.9 (CH=CH-CH-OTBDPS), 127.8 (CH=), undefined (interference with d₆-benzene signal) (4C, ArCH), 125.5 [CH=CH-(CH₂)₄-CH₃], 73.5 (CH–OTES), 69.8 (CH–OTBDPS), 50.7 (CH₃O), 37.9 [CH₂-(CH₂)₂-COO], 36.9 (CH₂-CH-OTES), 33.7 (CH2-COO), 31.7 (CH2-CH2-CH3), 29.5 $[CH_2-(CH_2)_2-CH_3], 27.7 [CH_2-(CH_2)_3-CH_3],$ 27.1[(CH₃)₃C], 22.8 (CH₂-CH₃), 20.6 (CH₂-CH₂-COO), 19.4 (C), 14.1 (CH₃), 7.0 [(CH₃-CH₂)₃Si], 5.3 [CH₃-CH₂)₃Si]. IR (film): 3071, 3048, 2957, 2929, 2875, 2857, 1742, 1460, 1428, 1260, 1110, 1076, 821, 801, 740, 702 cm⁻¹. HRMS (ESI) calcd for $C_{43}H_{66}O_4Si_2Na (M+Na)^+$: 725.4397, found $(M+Na)^+$: 725.4415.

1.1.32. Leukotriene B_4 (3). To a solution of 39 (15 mg, 0.021 mmol) in THF (2 ml) at 0°C was added a 1.0 M solution of TBAF in THF (0.4 ml, 0.40 mmol). After 15 h the reaction mixture was diluted with EtOAc (50 ml). Washing with brine, drying over Na₂SO₄ and concentrating afforded crude 3 that was purified by flash chromatography affording LTB₄ (**3**) (4.5 mg, 63%). $[\alpha]_D^{25}=12.6$ (*c*=0.1, CHCl₃) {Ref.^{18b} $[\alpha]_D^{20}=12.6$ (*c*=0.46, CDCl₃)}. UV (MeOH) λ_{max} 260, 270, 281 nm. ¹H NMR (CDCl₃, 300 MHz): δ 6.5 (1H, dd, J=13.5, 11.1 Hz, H-8), 6.4-6.2 (2H, m, H-9, H-10), 6.1 (1H, t, J=11.1 Hz, H-7), 5.8 (1H, dd, J=14.7, 6.3 Hz, H-11), 5.6–5.5 (1H, m, H-15), 5.5–5.3 (2H, m, H-14, H-6), 4.6 (1H, m, H-5), 4.2 (1H, m, H-12), 2.5-2.3 (4H, m, H-2, H-13), 2.1-2.0 (2H, m, H-16), 1.8-1.5 (4H, m, H-3, H-4), 1.4-1.2 (6H, m, H-17, H-18, H-19), 0.9 (3H, t, J=6.9 Hz, H-20). ¹³C NMR (CDCl₃, 75.5 MHz): δ 174.1 (C-1), 136.9 (C-11), 134.2, 134.0 (C-9, C-15), 133.7 (C-6), 130.3, 130.2 (C-7, C-10), 127.6 (C-8), 124.1 (C-14), 71.9 (C-12), 67.6 (C-5), 36.7 (C-4), 35.3 (C-13), 33.8 (C-2), 31.4 (C-18), 29.2 (C-17), 27.4 (C-16), 22.4 (C-19), 20.7 (C-3), 13.9 (C-20). HPLC/API-ES/MS (m/z): 335 $[M_{LTB4}-H^+]^-$.

Acknowledgements

Financial support of this research in part by Laboratorios Lasa S.A., Barcelona (Spain), and the Department of Cell Biology UMDNJ-SOM is gratefully acknowledged. We would like to thank Upjohn & Pharmacia for kindly providing us with standards of HETEs and leukotrienes for comparison.

References

 (a) Bergstrom, S. Science 1967, 157, 382–391. (b) Bergstrom, S. Angew. Chem., Int. Ed. Engl. 1983, 22, 858–866.
(c) Samuelsson, B. Science 1983, 220, 568–575. (d) Samuelsson, B. Angew. Chem., Int. Ed. Engl. 1983, 22, 805–815. (e) Serhan, C. N.; Hamberg, M.; Samuelsson, B. Proc. Nat. Acad. Sci. USA 1984, 81, 5335–5339. (f) Samuelsson, B.; Dahlen, S. E.; Lindgren, J. A.; Rouzer, C. A.; Serhan, C. N. *Science* **1987**, *237*, 1171–1176.

- Woollard, P. M. Biochem. Biophys. Res. Commun. 1986, 136, 169–176.
- 3. Hamberg, M.; Samuelsson, B. Proc. Natl. Acad. Sci. USA 1974, 71, 3400–3404.
- Dailey, L. A.; Imming, P. Curr. Med. Chem. 1999, 6 (5), 389– 398.
- 5. Ford-Hutchinson, A. W. Crit. Rev. Immunol. 1990, 10, 1-12.
- Gosselin, J.; Borgeat, P. US Patent 5,789,441 (Virocell Inc.), 1998.
- (a) Spur, B. W.; Crea, A.; Peters, W.; Konig, W. Arch. Pharm. (Weinheim) 1985, 318, 225–228. (b) Hanzawa, Y.; Kawagoe, K.; Inazawa, K.; Kobayashi, Y. Tetrahedron Lett. 1988, 29, 5665–5666. (c) Tanaka, Y.; Klauck, T. M.; Jubiz, W.; Taguchi, T.; Hanzawa, Y.; Igarashi, A.; Inazawa, K.; Kobayashi, Y.; Briggs, R. G. Arch. Biochem. Biophys. 1988, 263, 178–190.
- Corey, E. J.; Niwa, H.; Knolle, J. J. Am. Chem. Soc. 1978, 100, 1942–1943.
- (a) Mosset, P.; Pointeau, P.; Aubert, F.; Lellouche, J. P.; Beaucourt, J. P.; Gree, R. *Bull. Soc. Chim. Fr.* **1990**, 298– 315. (b) Yadagiri, P.; Lumin, S.; Mosset, P.; Capdevila, J.; Falck, J. R. *Tetrahedron Lett.* **1986**, 27, 6039–6040.
- (a) Russell, S. W.; Pabon, H. J. J. J. Chem. Soc., Perkin Trans. I 1982, 545–552. (b) Bestmann, H, J.; Pecher, B.; Riemer, C. Synthesis 1991, 731–734. (c) Le Merrer, Y.; Gravier, C.; Languin-Micas, D.; Depezay, J. C. Tetrahedron Lett. 1986, 27, 4161–4164. (d) Just, G.; Wang, Z. Y. Tetrahedron Lett. 1985, 26, 2993–2996. (e) Just, G.; Wang, Z. Y. J. Org. Chem. 1986, 51, 4796–4802. (f) Leblanc, Y.; Fitzsimmons, B. J.; Adams, J.; Perez, F.; Rokach, J. J. Org. Chem. 1986, 51, 789–793.
- (a) Nicolaou, K. C.; Ladduwahetty, T.; Taffer, I. M.; Zipkin, R. E. Synthesis **1986**, 344–347. (b) Nicolaou, K. C.; Ramphal, J. Y.; Abe, Y. Synthesis **1989**, 898–901. (c) Djuric, S. W.; Miyashiro, J. M.; Penning, T. D. Tetrahedron Lett. **1988**, 29, 3459–3462.
- (a) Kobayashi, Y.; Shimazaki, T.; Sato, F. *Tetrahedron Lett.* 1987, 28, 5849–5852. (b) Shimazaki, T.; Kobayashi, Y.; Sato, F. *Chem. Lett.* 1988, 1785–1788. (c) Yamaguchi, M.; Hirao, I. *Tetrahedron Lett.* 1983, 24, 391–394. (d) Chemin, D.; Gueugnot, S.; Linstrumelle, G. *Tetrahedron* 1992, 48, 4369–4378.
- 13. Han, X.; Corey, E. J. Org. Lett. 2000, 2 (16), 2543-2544.
- (a) Mori, Y.; Asai, M.; Okumura, A.; Furukawa, H. *Tetrahedron* **1995**, *51*, 5299–5314. (b) Hansen, R. H. *Chem. Rev.* **1991**, *91*, 437–475.
- (a) Rodríguez, A.; Nomen, M.; Spur, B. W.; Godfroid, J. J. *Tetrahedron Lett.* **1999**, 40, 5161–5164. (b) Tokunaga, M.; Larrow, J. F.; Kakiuchi, F.; Jacobsen, E. N. *Science* **1997**, 277, 936–938. (c) Furrow, M. E.; Schaus, S. E.; Jacobsen, E. N. *J. Org. Chem.* **1998**, 63, 6776–6777. (d) Jacobsen, E. N.; Kakiuchi F.; Konsler R. G.; Larrow, J. F.; Tokunaga, M. *Tetrahedron Lett.* **1997**, 38, 773–776. (e) Wu, M. H.; Jacobsen, E. N. *Tetrahedron Lett.* **1997**, 38, 1693–1696. (f) Brandes, B. D.; Jacobsen, E. N. *Tetrahedron: Asymmetry* **1997**, 8, 3927–3934. (g) Schaus, S. E.; Jacobsen, E. N. *Tetrahedron Lett.* **1996**, 37, 7937–7940. (h) Hansen, K. B.; Leighton, J. L.; Jacobsen, E. N. *J. Am. Chem. Soc.* **1996**, 118, 10924–10925. (i) Leighton, J. L.; Jacobsen, E. N. *J. Org. Chem.* **1996**, 61, 389–390. (j) Larrow, J. F.; Schaus, S. E.; Jacobsen, E. N. *J. Am. Chem. Soc.* **1996**, 118, 7420–7421.

- Nicolaou, K. C.; Zipkin, R. E.; Dolle, R. E.; Harris, B. D. J. Am. Chem. Soc. 1984, 106, 3548–3551.
- (a) Yamaguchi, M.; Hirao, I. *Tetrahedron Lett.* **1983**, 24, 391– 394. (b) Avigon-Tropis, M.; Treilhou, M.; Pougny, J. R.; Frechard-Ortuno, I.; Linstrumelle, G. *Tetrahedron* **1991**, 47, 7279–7286.
- Taffer, I. M.; Zipkin, R. E. Tetrahedron Lett. 1987, 28, 6543– 6544.
- (a) Maehr, H.; Perrotta, A.; Smallheer, J. J. Org. Chem. 1988, 53, 832–836. (b) Kerdesky, F. A. J.; Schmidt, S. P.; Brooks, D. W. J. Org. Chem. 1993, 58, 3516–3520.
- (a) Corey, E. J.; Marfat, A.; Goto, G.; Brion, F. J. Am. Chem. Soc. 1980, 102, 7984–7985. (b) Corey, E. J.; Marfat, A.; Munroe, J.; Kim, K. S.; Hopkins, P. B.; Brion, F. Tetrahedron Lett. 1981, 22, 1077–1080. (c) Corey, E. J.; Hopkins, P. B.; Barton, A. E.; Bangerter, B. Tetrahedron 1982, 38, 2653– 2657. (d) Sugiura, M.; Beierbeck, H.; Belanger, P. C.; Kotovych, G. J. Am. Chem. Soc. 1984, 106, 4021–4025. (e) Guindon, Y.; Zamboni, R.; Lau, C.-K.; Rokach, J. Tetrahedron Lett. 1982, 23, 739–742.